The impact of water released from boehmite nanoparticles during curing in epoxy-based nanocomposites

Supporting Information

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Content

- Figure S1 X-ray diffractogramms of dried and as received BNP.
- Figure S2 Full ATR-FTIR spectra of heated (80 °C, 30 min) anhydride hardener pure, with water, BNP or dBNP.
- Figure S3 ATR-FTIR spectra of heating series (80 °C) for pure anhydride hardener (AH).
- Figure S4 ATR-FTIR spectra of heating series (80 °C) for AH + H₂O.
- Figure S5 ATR-FTIR spectra of heating series (80 °C) for AH + BNP.
- Figure S6 ATR-FTIR spectra of heating series (80 °C) for AH + dBNP.
- Figure S7 ATR-FTIR spectra of heating series (80 °C) for pure amine accelerator (Acc).
- Figure S8 ATR-FTIR spectra of heating series (80 °C) for Acc + H₂O.
- Figure S9 ATR-FTIR spectra of heating series (80 °C) for Acc + BNP.
- Figure S10 ATR-FTIR spectra of heating series (80 °C) for pure Acc + dBNP.
- Figure S11 NIR spectra of heating (80 °C) for pure DGEBA.
- Figure S12 NIR spectra of heating (80 °C) for DGEBA with 30 wt.% water.
- Figure S13 NIR spectra of heating (80 °C) for DGEBA with 30 wt.% BNP.
- Figure S14 DRIFT spectra of cryomilled samples in KBr: cured epoxy polymer pure, nanocomposite with 15 wt.% BNP and pure BNP.

Powder X-ray diffraction (XRD)



Figure S1 X-ray diffractograms of dried and as received BNP.

XRD data shows that the crystalline structure of the dBNP is not affected by drying at 80 °C for 5 h.

FTIR spectra

The following spectra give an overview on the changes by heating at 80 °C of AH or Acc - pure, with H2O, with boehmite nanoparticles (BNP) or with dried BNP (dBNP). The measurements after 15 and 90 min have been recorded are not shown to keep the diagrams well-structured. They fit with the observable trends and do not contain further information.



Figure S2 Full ATR-FTIR spectra of heated (80 °C, 30 min) anhydride hardener pure (**black**), with water (**blue**), BNP (**red**) or dBNP (**yellow**). Suggesting a reaction between water released from the BNP and the AH Due to phase separation only the signals of AH and hydrolyzed AH are visible for AH + H_2O and no signals of water.



Figure S3 ATR-FTIR spectra of heating series (80 °C) for pure anhydride hardener (AH). No changes visible.



Figure S4 ATR-FTIR spectra of heating series (80 °C) for AH + H_2O . Hydrolysis of the AH is conclusive (see Figure 3). Additional signals appear in the carbonyl-region (1715 cm⁻¹) and for the hydroxyl-groups around 1440 cm⁻¹ and 3200 cm⁻¹. Due to phase separation only the signals of AH and hydrolyzed AH are visible and no signals of water.



Figure S5 ATR-FTIR spectra of heating series (80 °C) for AH + BNP. Hydrolysis of the AH by water released from the BNP is conclusive (see Figure 3). Additional signals are visible in the carbonyl-region (1715 cm⁻¹).



Figure S6 ATR-FTIR spectra of heating series (80 °C) for AH + dBNP. No changes visible.



Figure S7 ATR-FTIR spectra of heating series (80 °C) for pure amine accelerator (Acc). Minor changes appear due to reaction with humidity. (changed signals see table 2)



Figure S8 ATR-FTIR spectra of heating series (80 °C) for Acc + H_2O . The partly protonation of Acc by water is visible. (changed signals see table 2)



Figure S9 ATR-FTIR spectra of heating series (80 °C) for Acc + BNP. The partly protonation of Acc by water released from the BNP is visible. (changed signals see table 2)



Figure S10 ATR-FTIR spectra of heating series (80 °C) for pure Acc + dBNP. Minor changes appear due to reaction with humidity. (changed signals see table 2)

NIR spectroscopy on DGEBA

NIR-spectra have been recorded between 7500 cm⁻¹ and 4000 cm⁻¹. The epoxy groups which are most likely to react show characteristic absorption bands from 4552 to 4497 cm⁻¹ and 6149 to 6026 cm⁻¹. They are commonly used to observe the conversion of the DGEBA during curing reaction.^[1] Any influence of the BNP or water as mentioned for mid-infrared lead at least lead to a reduction of the peak intensities e.g. due to oligomerization or hydrolysis of the DGEBA.

[1] E. Duemichen, M. Javdanitehran, M. Erdmann, V. Trappe, H. Sturm, U. Braun, G. Ziegmann, *Thermochimica Acta* **2015**, *616*, 49–60.



Figure S11 NIR spectra of heating (80 °C) for pure DGEBA. No change of the signals observable. Only a baseline shift by heating from room temperature (RT, 22 °C) to 80 °C.



Figure S12 NIR spectra of heating (80 °C) for DGEBA with 30 wt.% water. No change of the signals observable. Only a baseline shift by heating from room temperature (RT, 22 °C) to 80 °C.



Figure S13 NIR spectra of heating (80 °C) for DGEBA with 30 wt.% BNP. No changes of the peaks observable. Only a baseline shift by heating from room temperature (RT, 22 °C) to 80 °C.

The following diagram shows the full DRIFT spectra of Ep and Ep + 15BNP. The significant change in the carbonyl region is depicted in Figure 6 of the main document.



Figure S14 DRIFT spectra of cryomilled samples in KBr: cured epoxy polymer pure (Ep, black), nanocomposite with 15 wt.% BNP (Ep + 15 BNP, red) and pure BNP (green). Obvious differences between Ep and Ep + 15BNP are related to the BNP. Most important difference is depicted in Figure 6.